

## Ni-Sn coated Ni foams – suitable cathodes for large-scale alkaline water electrolysis?

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Driven by the continuously increasing demand for electricity, diminishing carbon-based energy resources, steadily increasing ecological concerns, the focus has shifted to finding clean and renewable solutions. Electrochemical energy conversion and storage systems have taken the spotlight in shaping a greener future, with hydrogen, obtained through electrolysis, emerging as a leading energy carrier candidate. The scalability and abundance of materials used in alkaline water electrolysis, alongside the maturity of this industrial process, make it a favoured pathway for hydrogen production powered by renewables.

Electrodeposition was firstly investigated through linear sweep voltammetry and controlled potential coulometry techniques, in order to obtain high-performance Ni-Sn coated Ni foam. The objective was to achieve coatings of optimal composition to enable outstanding catalytic activity towards hydrogen evolution reaction, characterized by low overpotentials that are below -100 mV at current density of  $-1 \text{ A cm}^{-2}$ , normalized per geometric area, in 1 M KOH. Since these electrodes showed promising results, the electrodeposition was further investigated in galvanostatic setting in order to determine the most favourable conditions, as this technique is better suited for industrial-scale plating. Ni-Sn coated Ni foams, deposited at specific constant deposition current density, were additionally tested in zero-gap flow electrolysers where they have shown good stability and low value of overvoltage, which is in alignment with criteria proposed by IRENA<sup>2</sup>.

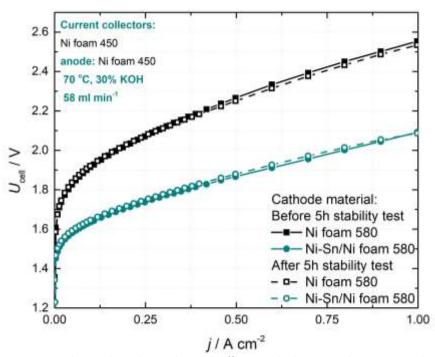


Figure 1. Cell voltage – current density dependency when two different cathodes are used – Ni-Sn coated one, and bare Ni foam

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